A GUAIANOLIDE FROM GUEVARIA SODIROI*

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Abstract—A new guaianolide has been isolated from *Guevaria sodiroi*. It has been assigned the trivial name guevariolide and its structure elucidated.

So far no chemical results on Guevaria sp. [1] (Compositae, tribe Eupatorieae) are available. This genus is placed in the somewhat arteficial Piqueria group [2], which consists of twenty genera. Only representatives of Ageratum, Piqueria and of the large genus Stevia, all placed in the same group, have been investigated up to now. From Ageratum species only p-hydroxyacetophenone derivatives have been identified [3] while from a Piqueria species derivatives of the unusual monoterpene carquejol [4, 5] are reported. The investigations of several Stevia species afforded mainly kaurene derivatives [6], a few labdanes [7], bisabolenes [8] and, as very common constituents, longipinene derivatives [8, 9]. However, a few sesquiterpene lactones were also isolated [10, 11].

The aerial parts of G. sodiroi (Hieron, in Sod.) K. et R. afforded, in addition to bisabolene, bisabolol, umbelliferone and lupeyl acetate, small amounts of the lactone 1, which we have named guevariolide. The structure was deduced from the very characteristic ¹H NMR data (Table 1). By spin decoupling at 400 MHz in deuteriobenzene all signals could be assigned. Irradiation of the four-fold doublet at δ 3.54 collapsed the typical H-13 doublets and the doublet at 3.86 to singlets, while a broadened double doublet at 5.26 was changed to a doublet. The latter was further coupled with a double doublet at 2.17 and a broadened doublet at 2.57, both

Table 1. ¹H NMR spectral data of compound 1 (TMS as internal standard)

	CDCl ₃	$C_6D_6^*$
H-3	3.56 s	3.10 s
H-6	4.38 d	3.86 d
H-7	4.03 dddd	3.54 <i>dddd</i>
H-8	5.67 br. dd	5.26 br. dd
H-9	3.24 br. d	2.57 br. d
H-9′	2.66 dd	2.17 dd
H-13	6.22 d	6.11 d
H-13'	5.48 d	5.14 d
H-14	2.25 s	2.19 s
H-15	1.79 s	1.57 s
OCOR	5.33 qq	5.59 gq
	$2.13 \ \hat{d}$	$2.08 \ \hat{d}$
	1.89 d	1.43 d
ОН	2.72 br. s	2.16 br. s

* 400 MHz;

J (Hz): 6, 7 = 10.5; 7, 8 = 2; 7, 13 = 3.5; 7, 13' = 3; 8, 9 \sim 1; 8, 9' = 6.5; 3', 4' = 3', 5' - 1.

showing a geminal coupling. This allowed the assignments of H-6 though H-9. The singlet at 3.10 was obviously due to an epoxide proton, while the methyl singlet at 2.19 must be an olefinic one that was deshielded by a keto group. Only two further signals were present: a broad singlet at 2.16(1 H) and a sharp singlet at 1.57(3 H). While the latter obviously had to be placed at the carbon bearing an epoxide, the former was the signal of a hydroxy group as was shown by deuterium exchange. This tertiary hydroxy group must be placed at C-5, as only a guaianolide of type 1 is in agreement with all the data. The stereochemistry at C-8 was deduced from the small coupling $J_{7,8}$, while the α -orientation of the hydroxyl at C-5 followed from the chemical shift of the 6-H signal, which should be more downfield in a lactone with a 5β hydroxy group. The stereochemistry of the epoxide was

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deduced from the presence of a hydrogen bond in the IR spectrum. Compound 1 is related to christinin, which has been isolated from a *Stevia* species [11]. This may be an indication of a relationship between the genus *Guevaria* and *Stevia*, though none of the other typical constituents of *Stevia* were isolated. Clearly further investigations are necessary.

EXPERIMENTAL

The air-dried plant material (collected in Ecuador, voucher RMK 7764A) was extracted with $\rm Et_2O$ -petrol (1:2). The resulting extracts were separated first by CC (Si gel) and further by TLC (Si gel). The aerial parts (50 g) afforded 10 mg bisabolene, 6 mg bisabolol, 15 mg lupeyl acetate, 5 mg umbelliferone and about 6 mg 1 ($\rm Et_2O$ -petrol, 1:1).

Guevariolide (I). Colourless gum, IR $v_{\text{max}}^{\text{CCL}}$ cm⁻¹: 3545 (OH-bonded), 1785 (y-lactone), 1720, 1625 (C=CC=O), 1720 and 1650 (C=CCO₂R); MS m/z (rel. int.): 374.137 (M⁺, 7), 274 (M - RCO₂H, 7), 256 (274 - H₂O, 10), 83 (C₄H₇CO⁺, 100), 55 (83 - CO, 49).

$$[\alpha]_{24^{\circ}}^{\lambda} = \frac{589}{-228} \frac{578}{-252} \frac{546}{-286} \frac{436 \,\mathrm{nm}}{-670} (c = 0.15).$$

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